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Key indicators

Single-crystal X-ray study T = 200 KMean σ (C–C) = 0.007 Å Disorder in solvent or counterion R factor = 0.058 wR factor = 0.155 Data-to-parameter ratio = 19.5

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trans-Bis(O-ethylxanthato)bis(triphenylphosphine)ruthenium(III) hexafluorophosphate monohydrate

In the monoclinic crystal structure of the title compound, trans-[Ru($C_3H_5OS_2$)₂($C_{18}H_{15}P$)₂]PF₆·H₂O, the structure of the Ru^{III} complex cation is very similar to that in the orthorhombic crystal of the nonhydrated complex [Noda, Ohuchi, Hashimoto, Fujiki, Itoh, Iwatsuki, Noda, Suzuki, Kashiwabara & Takagi (2006), *Inorg. Chem.* **45**, 1349–1355]. In the present crystal structure, the P–Ru–P bond axes of the complex cations are aligned parallel to the [101] direction.

Comment

Controlled-potential electrochemical oxidation of *cis*- $[Ru^{II}(S_2COEt)_2(PPh_3)_2]$ in acetone in the presence of NH₄PF₆ afforded the corresponding ruthenium(III) complex, *trans*- $[Ru^{III}(S_2COEt)_2(PPh_3)_2]PF_6$ (Noda *et al.*, 2006; Bag *et al.*, 1990). Recrystallization of the resulting green product by diffusion of diethyl ether vapour into a CH₂Cl₂ solution gave two forms of crystals, namely brown plates (minor component), (I), and green plates (major component), (II). We have already determined the structure of (II), which is an orthorhombic nonhydrated complex (Noda *et al.*, 2006). In this paper, we report the structure of (I), which is a monoclinic monohydrate, *trans*- $[Ru^{III}(S_2COEt)_2(PPh_3)_2]PF_6H_2O$.



In compound (I), the structure of the cationic complex (Fig. 1) is very similar to that in (II), except for the orientaion of the Et groups of the S₂COEt ligands. The Ru-P and Ru-S bond lengths, and the P-Ru-P bond angle (Table 1), compare well with those of (II) (Noda *et al.*, 2006).

The packing of (I) is illustrated in Fig. 2. There is positional disorder of the solvent water molecule. The P-Ru-P bond axis of the complex is aligned parallel to the [101] direction. This packing mode is different from that in the green orthorhombic crystal of (II).

Experimental

The preparation and analytical and spectroscopic characterization of the title Ru^{III} complex have been described previously by Noda *et al.* (2006).

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A plot of the cationic complex of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms have been omitted for clarity.



Figure 2

A packing diagram for (I). H atoms and the disordered water molecule have been omitted for clarity.

Crystal data

$[Ru(C_3H_5OS_2)_2(C_{18}H_{15}P)_2]PF_6$	$V = 4626.5 (5) \text{ Å}^3$
H ₂ O	Z = 4
$M_r = 1030.98$	$D_x = 1.48 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
a = 13.9877 (9) Å	$\mu = 0.69 \text{ mm}^{-1}$
b = 16.2122 (10) Å	T = 200 (2) K
c = 20.5103 (13) Å	Platelet, brown
$\beta = 95.904 \ (1)^{\circ}$	$0.30 \times 0.22 \times 0.02 \text{ mm}$

Data collection

Rigaku/MSC Mercury CCD areadetector diffractometer ω scans Absorption correction: multi-scan (*ABSCOR*; Higashi, 1999) *T*_{min} = 0.872, *T*_{max} = 0.999 45552 measured reflections 10523 independent reflections 8337 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.045$ $\theta_{\text{max}} = 27.5^{\circ}$

Refinement

$w = 1/[\sigma^2(F_o^2) + (0.0619P)]$
+ 6.0622P]
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} = 0.001$
$\Delta \rho_{\rm max} = 1.19 \text{ e} \text{ \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.69 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Ru1-S1	2.3626 (11)	Ru1-S7	2.3618 (11)
Ru1-S2	2.3821 (11)	Ru1-P1	2.4497 (11)
Ru1-S6	2.3740 (11)	Ru1-P2	2.4426 (10)
\$1-Ru1-\$2	72.96 (4)	S2-Ru1-P2	83.82 (4)
S1-Ru1-S6	177.27 (4)	S6-Ru1-S7	73.10 (4)
S1-Ru1-S7	104.56 (4)	S6-Ru1-P1	92.58 (4)
S1-Ru1-P1	85.89 (4)	S6-Ru1-P2	85.22 (4)
S1-Ru1-P2	96.40 (4)	S7-Ru1-P1	87.52 (4)
S2-Ru1-S6	109.45 (4)	S7-Ru1-P2	93.91 (4)
S2-Ru1-S7	176.37 (4)	P1-Ru1-P2	176.90 (4)
S2-Ru1-P1	94.88 (4)		

H atoms bonded to C atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C–H distances in the range 0.95–0.99 Å and with $U_{iso}(H) = 1.2U_{eq}(C)$. The solvent water molecule is disordered over two possible sites with occupancies of 50%. The water H atoms were located in difference syntheses and their positional parameters were fixed, with $U_{iso}(H) = 0.091 \text{ Å}^2$ and O–H distances in the range 0.64–0.84 Å. The volumes of the asymmetric units of (I) and (II) are 1156.6 (1) and 1107.5 (5) Å³, respectively. A check with the program *PLATON* (Spek, 2003) revealed that the crystal structure of (I) contains solvent-accessible voids of 49 Å³. However, no other solvent molecule was found in the difference map and the highest peak is located 0.93 Å from atom O51.

Data collection: *CrystalClear* (Rigaku, 2001); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MSC, 2004); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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References

Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). J. Appl. Cryst. 27, 435.

Bag, N., Lahiri, G. K. & Chakravorty, A. (1990). J. Chem. Soc. Dalton Trans. pp. 1557–1561.

- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Higashi, T. (1999). ABSCOR. Rigaku Corporation, Tokyo, Japan.
- Noda, K., Ohuchi, Y., Hashimoto, A., Fujiki, M., Itoh, S., Iwatsuki, S., Noda, T., Suzuki, T., Kashiwabara, K. & Takagi, H. D. (2006). *Inorg. Chem.* 45, 1349– 1355.
- Rigaku (2001). CrystalClear. Version 1.3. Rigaku Corporation, 3-9-12 Akishima, Tokyo, Japan.

Rigaku/MSC (2004). CrystalStructure. Version 3.6.0. Rigaku/MSC, 9009 New Trails Drive, The Woodlands, TX 77381-5209, USA.

Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany. Spek, A. L. (2003). J. Appl. Cryst. 36, 7–13.